

(SEM. III) THEORY EXAMINATION, 2015-16

MATERIAL SCIENCE

[Time : 3 hours]

[MaximumMarks : 100]

Section-A

- 1. Attempt all sections. All sections carry equal marks. Write answer of each section in short. $(10 \times 2=20)$
 - (a) What are "Miller indices"?
 - (b) Calculate atomic packing factor for HCP and FCC.
 - (c) Explain X-ray crystallography in short.
 - (d) What is heat treatment?
 - (e) What is primitive cell?
 - (f) Explain the term machinability.

- (g) What is plastic deformation?
- (h) Explain the term NDT.
- (i) Explain crystal and whiskers.
- (j) What are smart and Nano material?

Section-B

Note: Attempt any five questions from this section.

 $(10 \times 5 = 50)$

- 2. Write brief notes on following:
 - (i) Glass
 - (ii) Thermoplastic
 - (iii) Ceramic materials
- 3. Explain any two of the following:
 - (i) Edge dislocation
 - (ii) Twin boundary
 - (iii) Schottky's Defect
- 4. Explain lever rule. Why it is important in binary phase diagram?

- 5. Explain Plastics, its processing & applications.
 - What is the importance of engineering materials? Classify it.
- 7. Explain TTT diagram in brief. What information do you get from this diagram?
- 8. How Griffith's theory explains the mechanism of fracture for brittle material?
- 9. Sodium chloride crystal has FCC structure the density of NaCl is 2.18gm/cc, calculate the distance between two adjacent atoms. Atomic weight of sodium is 23 and that of chlorine is 35.5.

Section-C

Note: Attempt any two questions from this section

(2×15=30)

- 10. Write short note on:
 - (a) Composite material and its applications.
 - (b) Smart material and its applications.
 - (c) Corrosion and its prevention.

- 11. What do you understance by Imperfection in crystal structure and explain imperfections in the following?
 - (a) Lattice vibrations
 - (b) Point defects
 - (c) Surface or planar defects
- 12. Explain the following:
 - (a) Austempering and tempering
 - (b) Recrysrallization temperature
 - (c) P-type and N-type semiconductor

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